

09/763,324

(FILE 'HOME' ENTERED AT 16:00:18 ON 19 MAY 2002)

FILE 'CAPLUS, BIOSIS, MEDLINE' ENTERED AT 16:00:48 ON 19 MAY 2002

L1 8 S CHITOSAN? AND HOMOGENE? AND (ENZYME? OR TYROSINASE OR PHENOL(
L2 5 DUP REM L1 (3 DUPLICATES REMOVED)

=

Delacroix

09/763,324

FILE 'CPLUS' ENTERED AT 16:00:48 ON 19 MAY 2002
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FILE 'BIOSIS' ENTERED AT 16:00:48 ON 19 MAY 2002
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FILE 'MEDLINE' ENTERED AT 16:00:48 ON 19 MAY 2002

=> s chitosan? and homogene? and (enzyme? or tyrosinase or phenol(3a)oxidase? or polyphenol(3a)oxidase?) and phenol?

OR IS NOT A RECOGNIZED COMMAND

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=> s chitosan? and homogene? and (enzyme? or tyrosinase or phenol(3a)oxidase? or polyphenol(3a)oxidase?) and phenol?

L1 8 CHITOSAN? AND HOMOGENE? AND (ENZYME? OR TYROSINASE OR PHENOL(3A)
 OXIDASE? OR POLYPHENOL(3A) OXIDASE?) AND PHENOL?

=> dup rem l1

PROCESSING COMPLETED FOR L1

L2 5 DUP REM L1 (3 DUPLICATES REMOVED)

=> d 12 abs ibib kwic 1-5

L2 ANSWER 1 OF 5 CPLUS COPYRIGHT 2002 ACS

AB A variety of crosslinked polymers exist in nature and crosslinking confers important functional properties to biomaterials. In nature, crosslinking is generally achieved by the enzymic conversion of **phenols** into reactive intermediates (e.g. quinones or free radicals) that undergo subsequent non-enzymic reactions. Although enzymic crosslinking reactions offer interesting opportunities for conferring functionality, it has been difficult to control the poorly characterized non-enzymic reactions. We report the use of combinatorial approaches to learn how reactions of enzymically-generated quinones can be exploited for polysaccharide crosslinking and for the coupling of proteins to polysaccharides. Specifically we used **tyrosinase** to convert natural **phenols** into reactive quinones and examined the crosslinking of the amino-polysaccharide **chitosan** and the coupling of **chitosan** to various proteins. In studies on polymer crosslinking we performed **homogeneous** reactions and examined various **phenols** and reaction conditions. Screening was based on a rapid method to characterize the mech. properties of the enzymically crosslinked **chitosan** gels. For **chitosan**-protein coupling we used heterogeneous conditions and screened various **phenols** and reaction conditions to identify conditions that coupled protein to **chitosan** films while maintaining biol. activity.

ACCESSION NUMBER: 2001:197377 CPLUS
TITLE: Combinatorial approach to biopolymer coupling and crosslinking
AUTHOR(S): Payne, Gregory F.; Chen, Tianhong; Vazquez-Duhalt, Rafael; Bentley, William E.; Smith, Paul J.
CORPORATE SOURCE: Center for Agricultural Biotechnology, University of Maryland, College Park, MD, 20742-4450, USA

SOURCE: Abstr. Pap. - Am. Chem. Soc. (2001), 221st, BIOT-070
 CODEN: ACSRAL; ISSN: 0065-7727

PUBLISHER: American Chemical Society
 DOCUMENT TYPE: Journal; Meeting Abstract
 LANGUAGE: English

AB A variety of crosslinked polymers exist in nature and crosslinking confers important functional properties to biomaterials. In nature, crosslinking is generally achieved by the enzymic conversion of **phenols** into reactive intermediates (e.g. quinones or free radicals) that undergo subsequent non-enzymic reactions. Although enzymic crosslinking reactions offer interesting opportunities for conferring functionality, it has been difficult to control the poorly characterized non-enzymic reactions. We report the use of combinatorial approaches to learn how reactions of enzymically-generated quinones can be exploited for polysaccharide crosslinking and for the coupling of proteins to polysaccharides. Specifically we used **tyrosinase** to convert natural **phenols** into reactive quinones and examined the crosslinking of the amino-polysaccharide **chitosan** and the coupling of **chitosan** to various proteins. In studies on polymer crosslinking we performed **homogeneous** reactions and examined various **phenols** and reaction conditions. Screening was based on a rapid method to characterize the mech. properties of the enzymically crosslinked **chitosan** gels. For **chitosan**-protein coupling we used heterogeneous conditions and screened various **phenols** and reaction conditions to identify conditions that coupled protein to **chitosan** films while maintaining biol. activity.

L2 ANSWER 2 OF 5 CAPLUS COPYRIGHT 2002 ACS

AB A **homogeneous**-phase **enzyme**-catalyzed process for producing modified **chitosan** polymers or oligomers comprises reacting an **enzyme**, e.g., **tyrosinase**, with a **phenolic substrate**, e.g., chlorogenic acid, in the presence of a **chitosan** polymer or oligomer. The modified **chitosan** polymers or oligomers produced by the novel processes, in particular those having useful functional properties, e.g., base solv. and/or high viscosity are also claimed.

ACCESSION NUMBER: 2000:144914 CAPLUS

DOCUMENT NUMBER: 132:182264

TITLE: Modified **chitosan** polymers and enzymic methods for their production

INVENTOR(S): Kumar, Guneet; Payne, Gregory F.

PATENT ASSIGNEE(S): USA

SOURCE: PCT Int. Appl., 47 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2000011038	A1	20000302	WO 1999-US19106	19990820
W:	AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM			

RW: GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
AU 9957814 A1 20000314 AU 1999-57814 19990820
EP 1137673 A1 20011004 EP 1999-945134 19990820
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO
PRIORITY APPLN. INFO.: US 1998-97709P P 19980821
WO 1999-US19106 W 19990820

REFERENCE COUNT: 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

TI Modified chitosan polymers and enzymic methods for their production
AB A homogeneous-phase enzyme-catalyzed process for producing modified chitosan polymers or oligomers comprises reacting an enzyme, e.g., tyrosinase, with a phenolic substrate, e.g., chlorogenic acid, in the presence of a chitosan polymer or oligomer. The modified chitosan polymers or oligomers produced by the novel processes, in particular those having useful functional properties, e.g., base solv. and/or high viscosity are also claimed.
ST chitosan reaction oxidized phenol enzyme oxidant; tyrosinase oxygen oxidn chlorogenic acid chitosan modification
IT Oxidation (enzymic, of phenols; enzymic methods for the manuf. of chitosan polymers modified with oxidized phenols)
IT Phenols, preparation RL: IMF (Industrial manufacture); PREP (Preparation) (reaction products; enzymic methods for the manuf. of chitosan polymers modified with oxidized phenols)
IT 9002-10-2, Tyrosinase RL: BPR (Biological process); BSU (Biological study, unclassified); BIOL (Biological study); PROC (Process) (enzymic methods for the manuf. of chitosan polymers modified with oxidized phenols)
IT 51-61-6DP, Dopamine, oxidized, reaction products with chitosan 106-44-5DP, p-Cresol, oxidized, reaction products with chitosan 120-80-9DP, Catechol, oxidized, reaction products with chitosan 327-97-9DP, Chlorogenic acid, oxidized, reaction products with chitosan 9012-76-4DP, Chitosan, reaction products with oxidized phenols RL: IMF (Industrial manufacture); PREP (Preparation) (enzymic methods for the manuf. of chitosan polymers modified with oxidized phenols)
IT 7782-44-7, Oxygen, uses RL: NUU (Other use, unclassified); USES (Uses) (enzymic methods for the manuf. of chitosan polymers modified with oxidized phenols)
L2 ANSWER 3 OF 5 CAPLUS COPYRIGHT 2002 ACS DUPLICATE 1
AB An enzymic method to graft hexyloxyphenol onto the biopolymer chitosan was studied. The method employs tyrosinase to convert the phenol into a reactive o-quinone, which undergoes subsequent nonenzymic reaction with chitosan. Reactions were conducted under heterogeneous conditions using chitosan films and also under homogeneous conditions using aq. methanolic mixts. capable of dissolving both hexyloxyphenol and chitosan.

Tyrosinase was shown to catalyze the oxidn. of hexyloxyphenol in such aq. methanolic solns. Chem. evidence for covalent grafting onto **chitosan** was provided by three independent spectroscopic approaches. Specifically, enzymic modification resulted in (1) the appearance of broad absorbance in the 350-nm region of the UV/vis spectra for **chitosan** films; (2) changes in the NH bending and stretching regions of **chitosan**'s IR spectra; and (3) a base-sol. material with 1H-NMR signals characteristic of both **chitosan** and the alkyl groups of hexyloxyphenol. Hexyloxyphenol modification resulted in dramatic changes in **chitosan**'s functional properties. On the basis of contact angle measurements, heterogeneous modification of a **chitosan** film yielded a hydrophobic surface.

Homogeneously modified chitosan offered rheol.

properties characteristic of assocg. water-sol. polymers.

ACCESSION NUMBER: 2000:816746 CAPLUS
 DOCUMENT NUMBER: 134:99633
 TITLE: Enzymatic grafting of hexyloxyphenol onto **chitosan** to alter surface and rheological properties
 AUTHOR(S): Chen, Tianhong; Kumar, Guneet; Harris, Michael T.; Smith, Paul J.; Payne, Gregory F.
 CORPORATE SOURCE: Center for Agricultural Biotechnology, University of Maryland, College Park, MD, 20742, USA
 SOURCE: Biotechnology and Bioengineering (2000), 70(5), 564-573
 CODEN: BIBIAU; ISSN: 0006-3592
 PUBLISHER: John Wiley & Sons, Inc.
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 REFERENCE COUNT: 56 THERE ARE 56 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

TI Enzymatic grafting of hexyloxyphenol onto **chitosan** to alter surface and rheological properties
 AB An enzymic method to graft hexyloxyphenol onto the biopolymer **chitosan** was studied. The method employs **tyrosinase** to convert the **phenol** into a reactive o-quinone, which undergoes subsequent nonenzymic reaction with **chitosan**. Reactions were conducted under heterogeneous conditions using **chitosan** films and also under **homogeneous** conditions using aq. methanolic mixts. capable of dissolving both hexyloxyphenol and **chitosan**. **Tyrosinase** was shown to catalyze the oxidn. of hexyloxyphenol in such aq. methanolic solns. Chem. evidence for covalent grafting onto **chitosan** was provided by three independent spectroscopic approaches. Specifically, enzymic modification resulted in (1) the appearance of broad absorbance in the 350-nm region of the UV/vis spectra for **chitosan** films; (2) changes in the NH bending and stretching regions of **chitosan**'s IR spectra; and (3) a base-sol. material with 1H-NMR signals characteristic of both **chitosan** and the alkyl groups of hexyloxyphenol. Hexyloxyphenol modification resulted in dramatic changes in **chitosan**'s functional properties. On the basis of contact angle measurements, heterogeneous modification of a **chitosan** film yielded a hydrophobic surface.
Homogeneously modified chitosan offered rheol.
 properties characteristic of assocg. water-sol. polymers.

ST **tyrosinase** grafting hexyloxyphenol **chitosan**
 IT Contact angle
 Viscosity
 (enzymic grafting of hexyloxyphenol onto **chitosan** to alter

surface and rheol. properties)

IT Oxidation
 (enzymic; enzymic grafting of hexyloxyphenol onto **chitosan** to alter surface and rheol. properties)

IT Polymers, preparation
 RL: BMF (Bioindustrial manufacture); BPN (Biosynthetic preparation); PRP (Properties); PUR (Purification or recovery); BIOL (Biological study); PREP (Preparation)
 (graft; enzymic grafting of hexyloxyphenol onto **chitosan** to alter surface and rheol. properties)

IT 9012-76-4DP, **Chitosan**, graft copolymer with hexyloxyphenol
 RL: BMF (Bioindustrial manufacture); BPN (Biosynthetic preparation); PRP (Properties); PUR (Purification or recovery); BIOL (Biological study); PREP (Preparation)
 (enzymic grafting of hexyloxyphenol onto **chitosan** to alter surface and rheol. properties)

IT 9002-10-2, **Tyrosinase**
 RL: BPR (Biological process); BSU (Biological study, unclassified); CAT (Catalyst use); BIOL (Biological study); PROC (Process); USES (Uses)
 (enzymic grafting of hexyloxyphenol onto **chitosan** to alter surface and rheol. properties)

IT 622-62-8 9012-76-4, **Chitosan** 18979-55-0, 4-n-Hexyloxyphenol
 26638-03-9, Methoxyphenol
 RL: BPR (Biological process); BSU (Biological study, unclassified); RCT (Reactant); BIOL (Biological study); PROC (Process); RACT (Reactant or reagent)
 (enzymic grafting of hexyloxyphenol onto **chitosan** to alter surface and rheol. properties)

IT 320401-59-0
 RL: BSU (Biological study, unclassified); MFM (Metabolic formation); BIOL (Biological study); FORM (Formation, nonpreparative)
 (enzymic grafting of hexyloxyphenol onto **chitosan** to alter surface and rheol. properties)

IT 69818-23-1 320401-57-8
 RL: BSU (Biological study, unclassified); MFM (Metabolic formation); RCT (Reactant); BIOL (Biological study); FORM (Formation, nonpreparative); RACT (Reactant or reagent)
 (enzymic grafting of hexyloxyphenol onto **chitosan** to alter surface and rheol. properties)

L2 ANSWER 4 OF 5 CAPLUS COPYRIGHT 2002 ACS

AB It was obsd. that addn. of **tyrosinase** and the simple phenol, p-cresol, to semi-dil. solns. of **chitosan** (I) resulted in the in situ formation of I gels. Specifically, homogeneous reactions were conducted with I solns. (0.32 w/v %) at pH near 6.0 and with cresol levels of 0.6 molar equiv (relative to I amino groups). Oscillatory shear measurements showed that the enzymic reaction resulted in large increases in the complex viscosity (η_{av}) and storage and loss moduli (G' and G''). These dynamic measurements indicated that the enzymic reaction resulted in the conversion of the nearly Newtonian semi-dil. I solns. into gels. The rheol. behavior of these enzymically-generated gels was compared to the behavior of acidic I solns. and to solns. contg. xanthan gum.

ACCESSION NUMBER: 2000:450035 CAPLUS

DOCUMENT NUMBER: 134:6082

TITLE: In situ chitosan gelation using the enzyme tyrosinase

AUTHOR(S): Kumar, G.; Bristow, J. F.; Smith, P. J.; Payne, G. F.

CORPORATE SOURCE: Center for Agricultural Biotechnology, Univ. Maryland,
 College Park, MD, 20742, USA
 SOURCE: Advances in Chitin Science (2000), 4 (EUCHIS'99),
 345-348
 CODEN: ACSCFF
 PUBLISHER: Universitaet Potsdam, Universitaetsbibliothek
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 REFERENCE COUNT: 18 THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS
 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

- TI In situ chitosan gelation using the enzyme tyrosinase
- AB It was obsd. that addn. of tyrosinase and the simple phenol, p-cresol, to semi-dil. solns. of chitosan (I) resulted in the in situ formation of I gels. Specifically, homogeneous reactions were conducted with I solns. (0.32 w/v %) at pH near 6.0 and with cresol levels of 0.6 molar equiv (relative to I amino groups). Oscillatory shear measurements showed that the enzymic reaction resulted in large increases in the complex viscosity (.eta.*) and storage and loss moduli (G' and G''). These dynamic measurements indicated that the enzymic reaction resulted in the conversion of the nearly Newtonian semi-dil. I solns. into gels. The rheol. behavior of these enzymically-generated gels was compared to the behavior of acidic I solns. and to solns. contg. xanthan gum.
- ST tyrosinase enzyme cresol in situ gelation
 chitosan soln; mech loss viscoelasticity viscosity
 chitosan soln gelation cresol enzyme
- IT Gelation
 Mechanical loss
 Viscoelasticity
 Viscosity
 (in situ chitosan soln. gelation using tyrosinase enzyme and p-cresol)
- IT Enzymes, uses
 RL: NUU (Other use, unclassified); USES (Uses)
 (tyrosinase; in situ chitosan soln. gelation using tyrosinase enzyme and p-cresol)
- IT 106-44-5, p-Cresol, uses 9002-10-2, Tyrosinase
 RL: NUU (Other use, unclassified); USES (Uses)
 (in situ chitosan soln. gelation using tyrosinase enzyme and p-cresol)
- IT 9012-76-4, Chitosan
 RL: PEP (Physical, engineering or chemical process); PRP (Properties);
 PROC (Process)
 (in situ chitosan soln. gelation using tyrosinase enzyme and p-cresol)
- L2 ANSWER 5 OF 5 CAPLUS COPYRIGHT 2002 ACS DUPLICATE 2
- AB Chitosan (I) is a natural biopolymer whose rich amine functionality confers water soly. at low pH. At higher pH's (>6.5), the amines are deprotonated, and I is insol. To attain water soly. under basic conditions the hydrophilic compd. chlorogenic acid (II) was enzymically grafted onto I. Despite its name, II is a non-chlorinated phenolic natural product that has carboxylic acid and OH functionality. The enzyme used was tyrosinase, which converts a wide range of phenolic substrates into electrophilic o-quinones. The o-quinones are freely diffusible and can undergo reaction with the nucleophilic amino groups of I. Using slightly acidic conditions

(pH = 6.0), it was possible to modify I under **homogeneous** conditions. When the amt. of II used in the modification reaction was >30% relative to the I amino groups, the modified I was obsd. to be sol. under both acidic and basic conditions, and to have a pH window of insol. at near neutral pH. Proton NMR spectra confirmed that I was chem. modified, although the degree of modification was low.

ACCESSION NUMBER: 1999:131727 CAPLUS
 DOCUMENT NUMBER: 130:239087
 TITLE: Enzymic grafting of a natural product onto **chitosan** to confer water solubility under basic conditions
 AUTHOR(S): Kumar, Guneet; Smith, Paul J.; Payne, Gregory F.
 CORPORATE SOURCE: Center for Agricultural Biotechnology, University of Maryland, College Park, MD, 20742, USA
 SOURCE: Biotechnology and Bioengineering (1999), 63 (2), 154-165
 CODEN: BIBIAU; ISSN: 0006-3592
 PUBLISHER: John Wiley & Sons, Inc.
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 REFERENCE COUNT: 56 THERE ARE 56 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

TI Enzymic grafting of a natural product onto **chitosan** to confer water solubility under basic conditions
 AB **Chitosan** (I) is a natural biopolymer whose rich amine functionality confers water soly. at low pH. At higher pH's (>6.5), the amines are deprotonated, and I is insol. To attain water soly. under basic conditions the hydrophilic compd. chlorogenic acid (II) was enzymically grafted onto I. Despite its name, II is a non-chlorinated phenolic natural product that has carboxylic acid and OH functionality. The enzyme used was **tyrosinase**, which converts a wide range of phenolic substrates into electrophilic o-quinones. The o-quinones are freely diffusible and can undergo reaction with the nucleophilic amino groups of I. Using slightly acidic conditions (pH = 6.0), it was possible to modify I under **homogeneous** conditions. When the amt. of II used in the modification reaction was >30% relative to the I amino groups, the modified I was obsd. to be sol. under both acidic and basic conditions, and to have a pH window of insol. at near neutral pH. Proton NMR spectra confirmed that I was chem. modified, although the degree of modification was low.
 ST chlorogenic acid grafting **chitosan** soly **tyrosinase**
 enzyme catalyst
 IT Solubility
 (alk.; enzymically catalyzed grafting of natural products onto **chitosan** to confer water soly. under alk. conditions)
 IT Enzymes, uses
 RL: CAT (Catalyst use); USES (Uses)
 (enzymically catalyzed grafting of natural products onto **chitosan** to confer water soly. under alk.. conditions)
 IT Polymerization
 Polymerization catalysts
 (graf; enzymically catalyzed grafting of natural products onto **chitosan** to confer water soly. under alk. conditions)
 IT 9002-10-2, **Tyrosinase**
 RL: CAT (Catalyst use); USES (Uses)
 (enzymically catalyzed grafting of natural products onto **chitosan** to confer water soly. under alk. conditions)
 IT 327-97-9, Chlorogenic acid 9012-76-4, **Chitosan**

09/763,324

RL: RCT (Reactant); RACT (Reactant or reagent)
(enzymically catalyzed grafting of natural products onto
chitosan to confer water solv. under alk. conditions)

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L Number	Hits	Search Text	DB	Time stamp
1	12	chitosan\$2 same (polymer\$2 or copolymer\$2) same viscosity same (cps! or centipoise\$2 or poise\$2)	USPAT; US-PGPUB	2002/05/19 15:46